

Phase Transitions in LaCoO_3 and LaGaO_3 based perovskites

- Alan Nicholls, University of Illinois at Chicago, IL
- Miladin Radovic, Oak Ridge National Laboratory, TN
- Edgar Lara-Curcio, Oak Ridge National Laboratory, TN
- Andrew Payzant, Oak Ridge National Laboratory, TN
- Mike Reece, Queen Mary University of London, UK

Students:

- John Lloyd, Laura Fegely, David Steimentz, Siddhartha Pathak

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Objectives

- Characterize the defect structure of LaCoO_3 based rhombohedral perovskites
- Demonstrate a ferroelastic to paraelastic phase transition in $\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$ by *in-situ* high temperature TEM
- Investigate *in-situ* microstructural changes during heating, annealing and cooling in the microscope column
- Study the structural stability of LaCoO_3 in reducing environment by XRD
- Demonstrate the temperature induced orthorhombic to rhombohedral LaGaO_3 phase transition

Experimental

- JEOL JEM-3010 TEM
- Gatan Model 652 double tilt heating stage was used to heat perovskite foils up to 850°C at a controlled heating rate of 10°C or 20 C per minute.
- Oxygen partial pressure in the microscope column is about 5×10^{-8} Pa
- Scintag PADV diffractometer (CuK α radiation)
- Thermal Mechanical Analyzer (TMA)
- Optical microscope

Lanthanum based cobaltites



$T_{\text{C} \rightarrow \text{R}} = \sim 1600^\circ\text{C}$; $T_{\text{C} \rightarrow \text{R}} = 950^\circ\text{C}$; $T_{\text{C} \rightarrow \text{R}} = 700^\circ\text{C}$ cubic

- Grain size - 3-5 μm
- Polycrystalline ceramics prepared by Praxair Surface Technologies

High temperature

Cubic structure with the space group $Pm\bar{3}m$

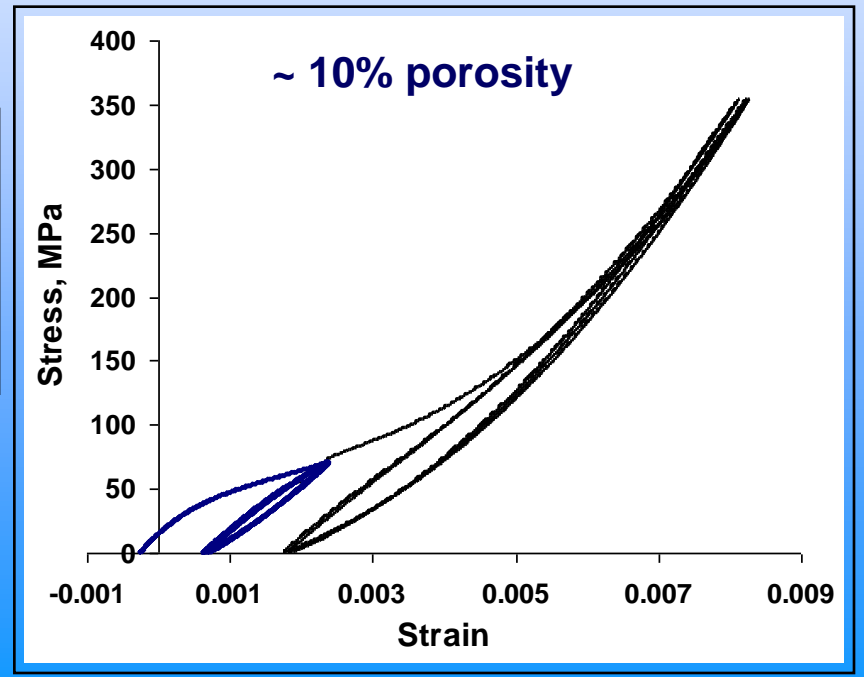
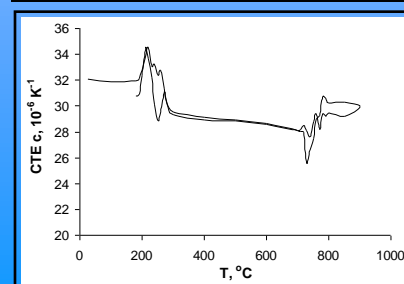
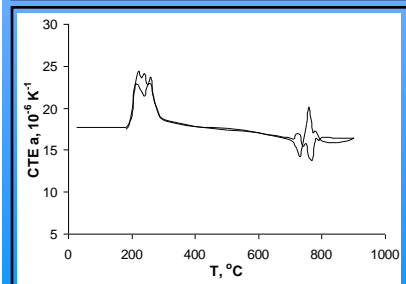
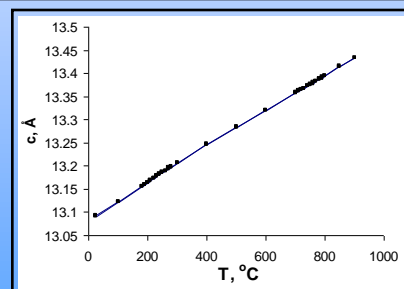
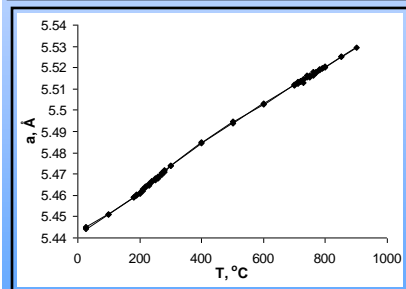
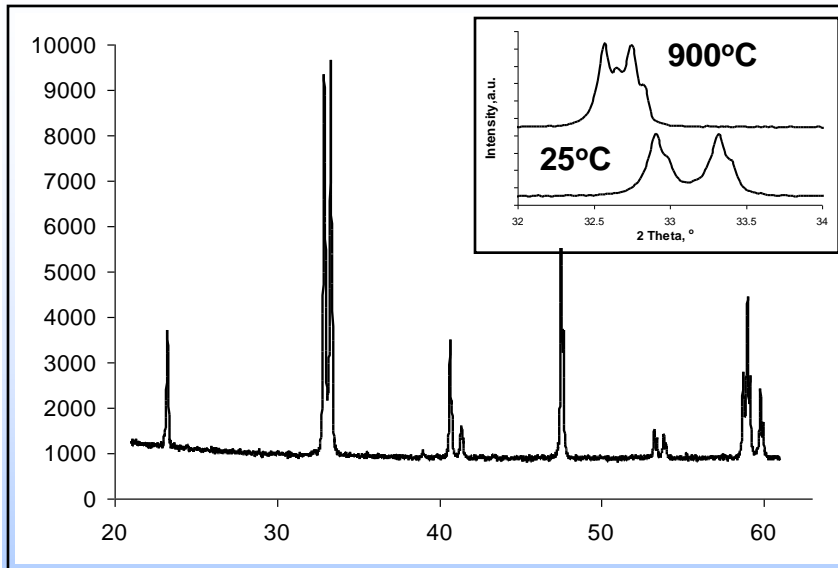
Upon cooling, LaCoO_3 exhibits a displacive phase transition at T_c

Room temperature

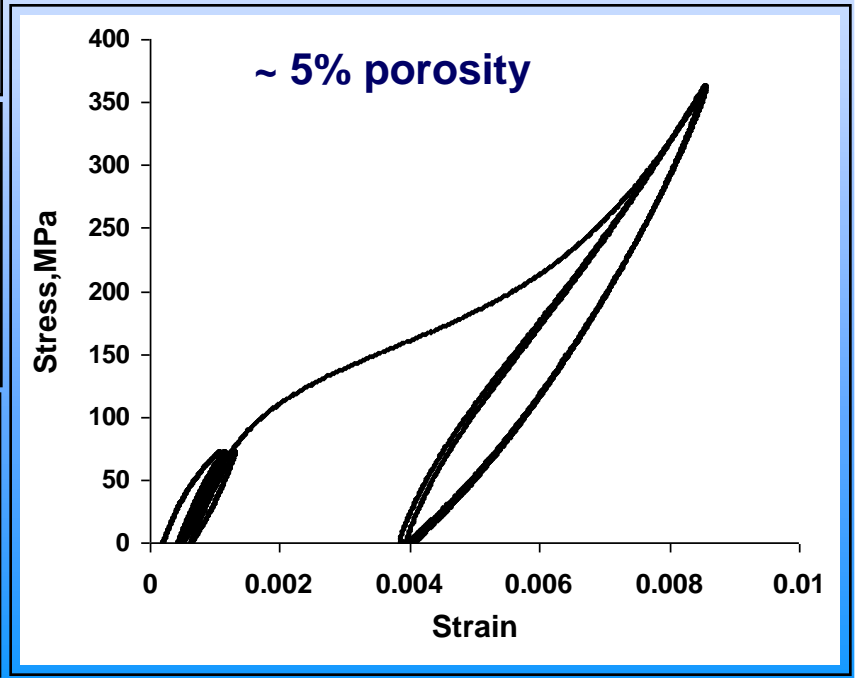
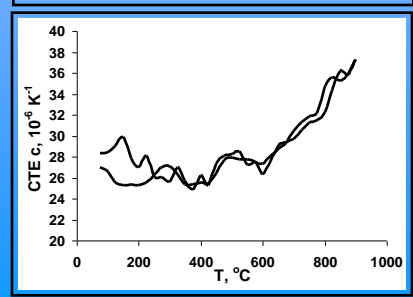
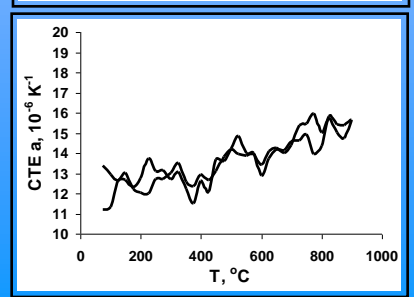
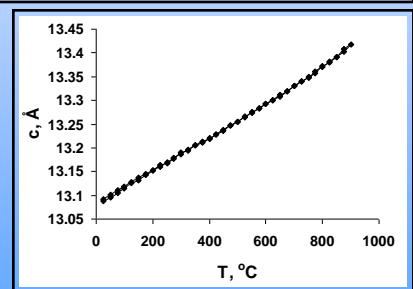
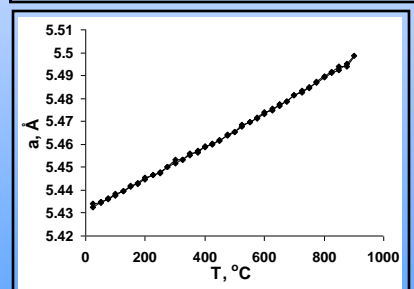
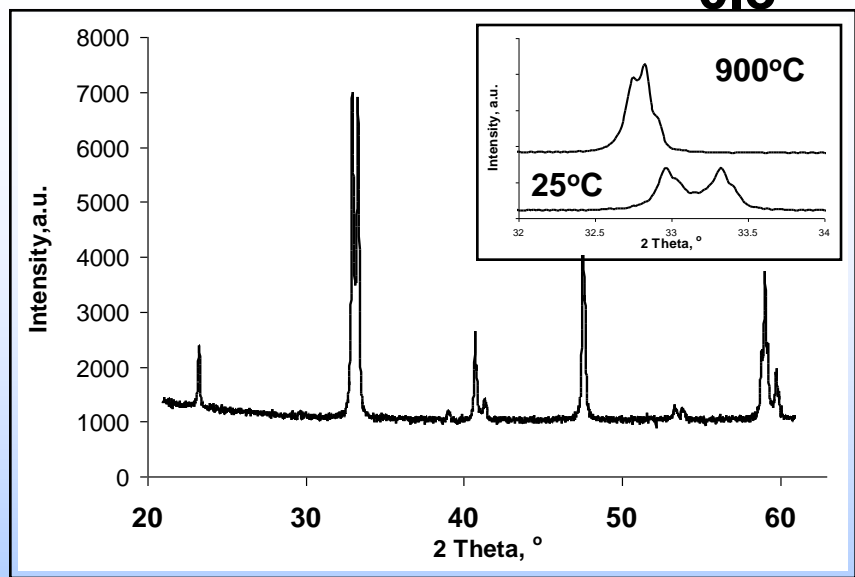
Rhombohedral structure with the space group $R\bar{3}c$

T_c is decreases with a divalent cation doping

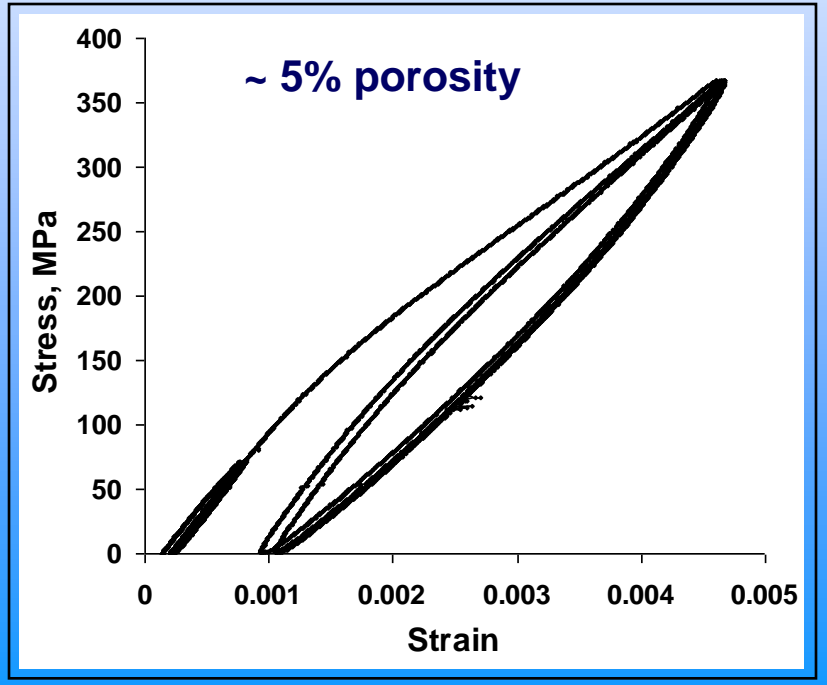
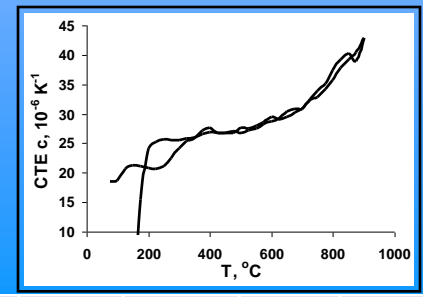
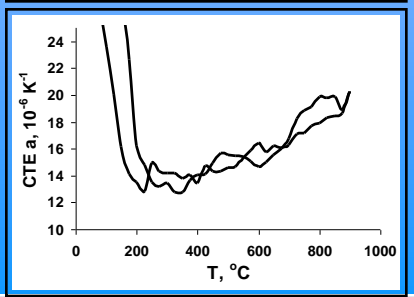
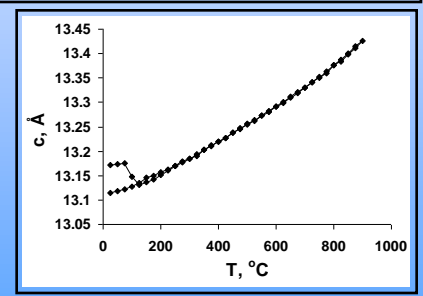
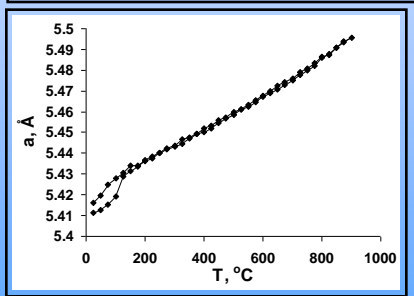
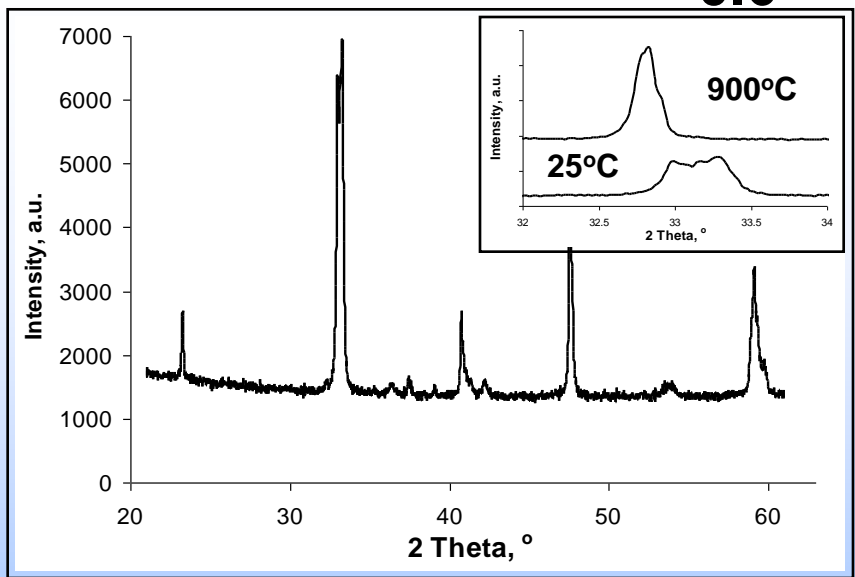
LaCoO₃



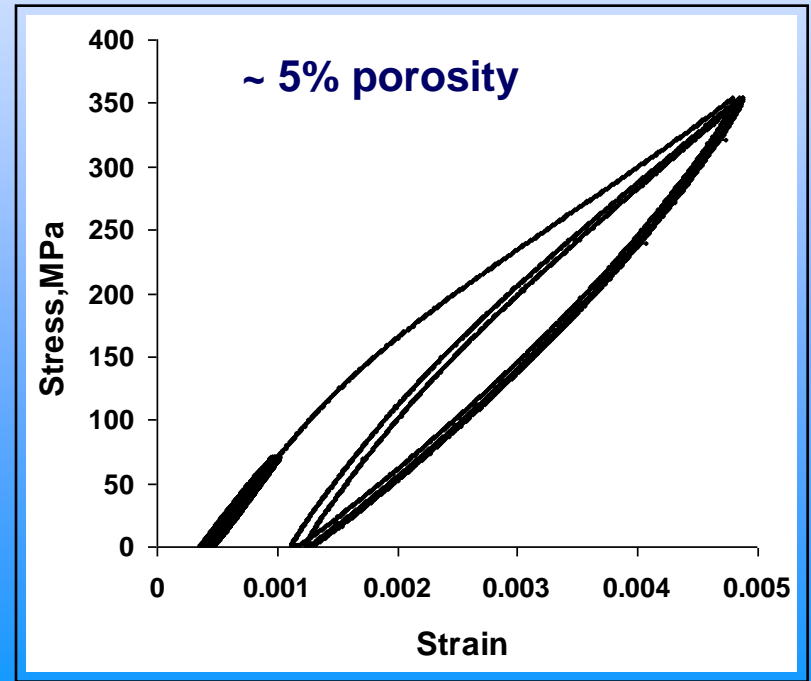
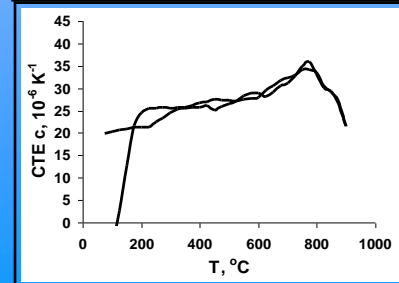
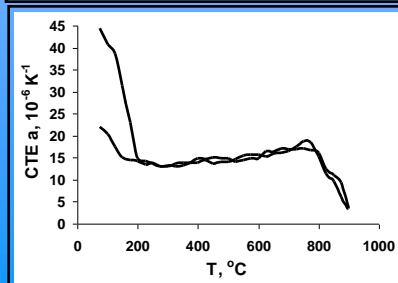
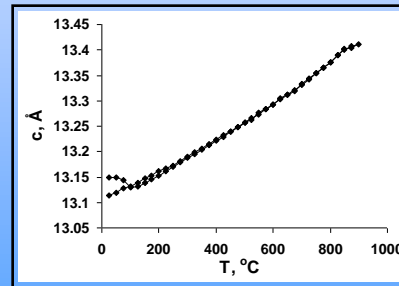
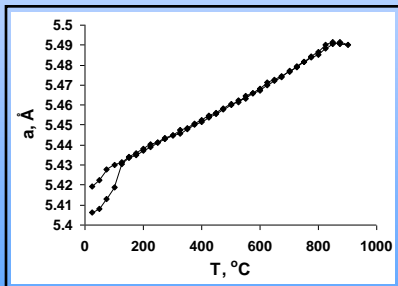
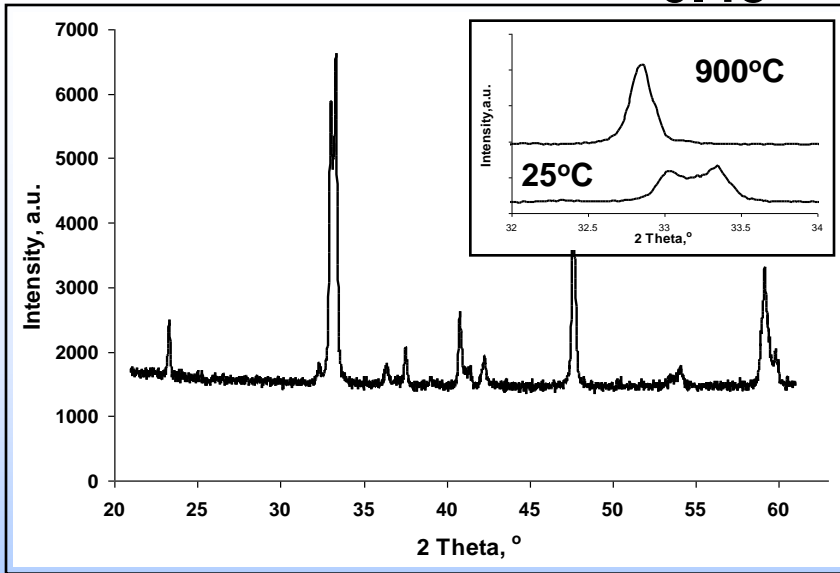
La_{0.8}Ca_{0.2}CoO₃



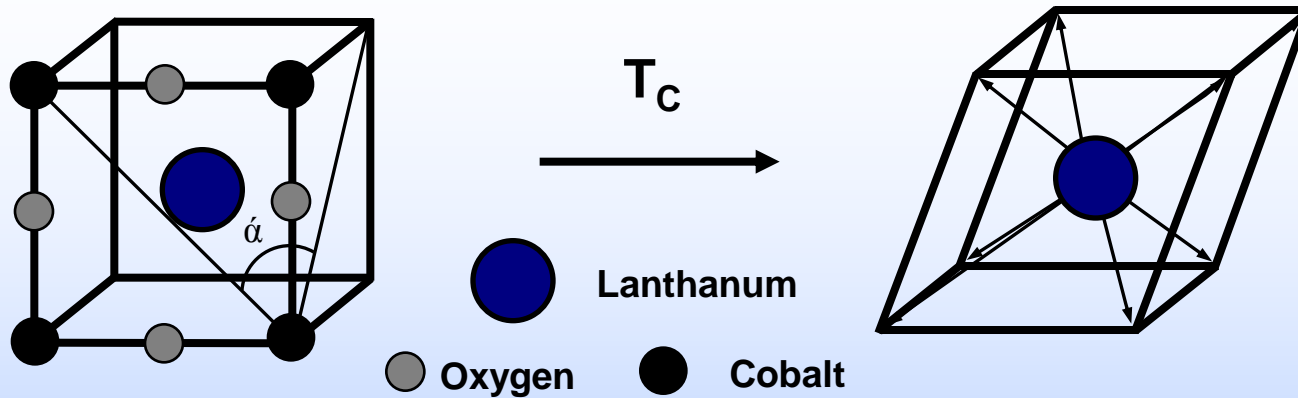
$\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$



$\text{La}_{0.45}\text{Ca}_{0.55}\text{CoO}_3$



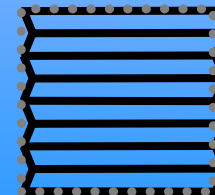
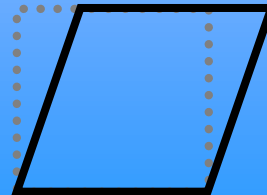
Ferroelastic Phase Transition



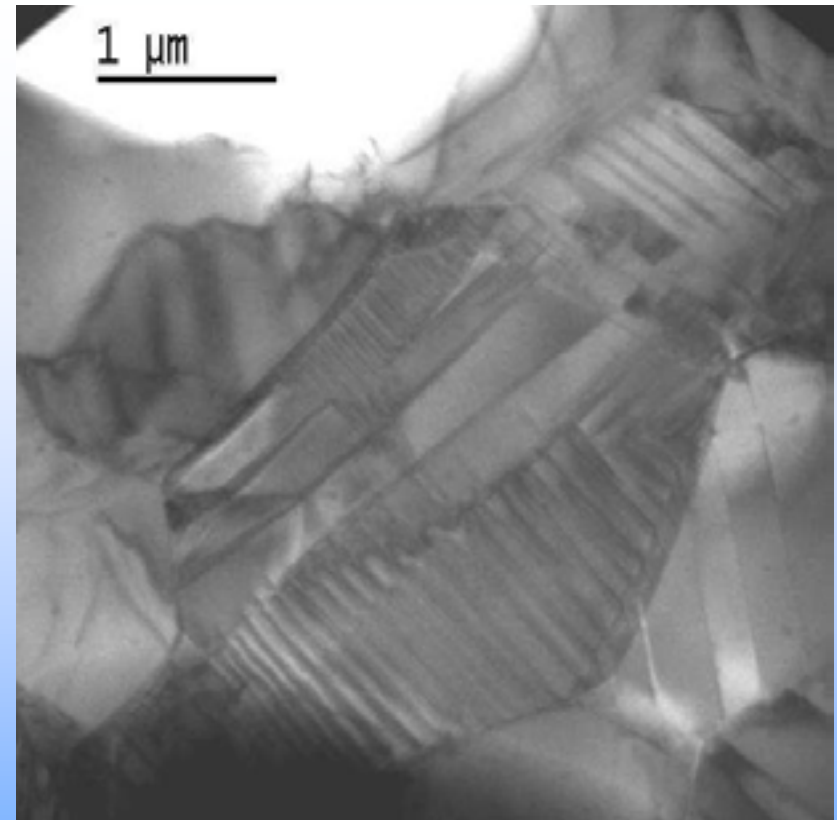
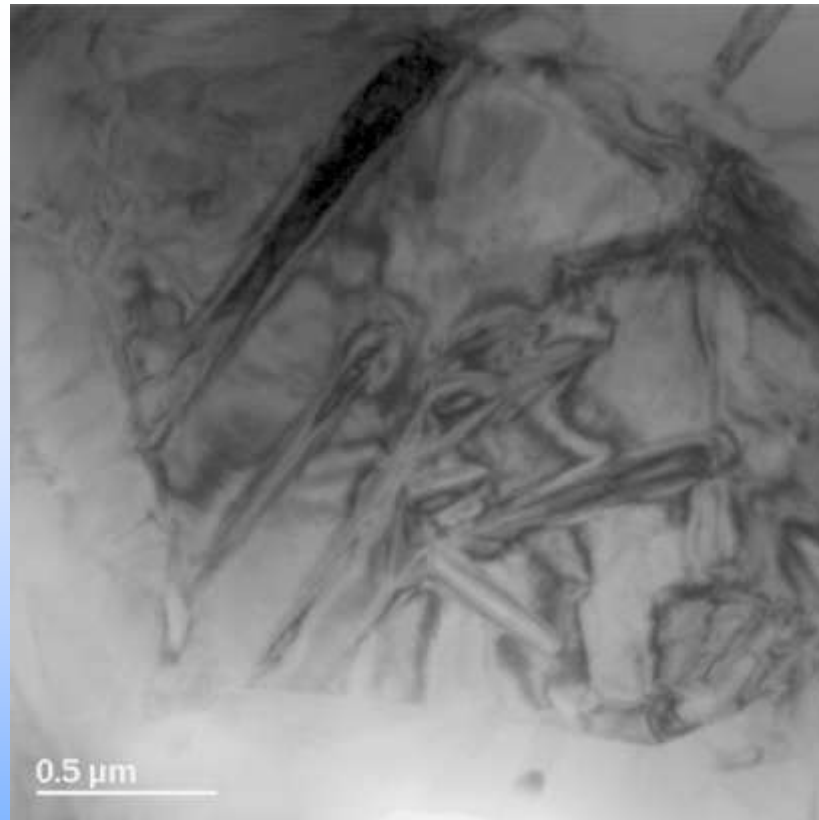
$T_{C \rightarrow R} = \text{LaCoO}_3$ at $\sim 1600^\circ\text{C}$; $\text{La}_{0.8}\text{Ca}_{0.2}\text{CoO}_3$ at 950°C ; $\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$ at 700°C

High Temperature
High symmetry prototypic phase
 $a = b = c$; $\alpha = \beta = \gamma = 90^\circ$
 $\alpha = 60^\circ$
Cubic

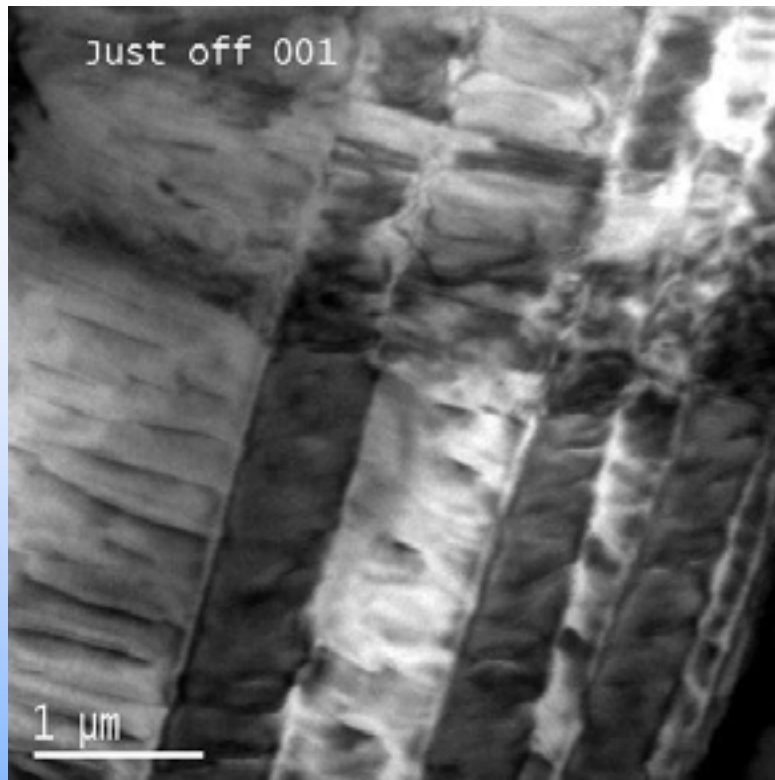
Low Temperature
Low symmetry phase
 $a = b = c$; $\alpha = \beta = \gamma \neq 90^\circ$
 $\alpha = 60.78^\circ$
Rhombohedral



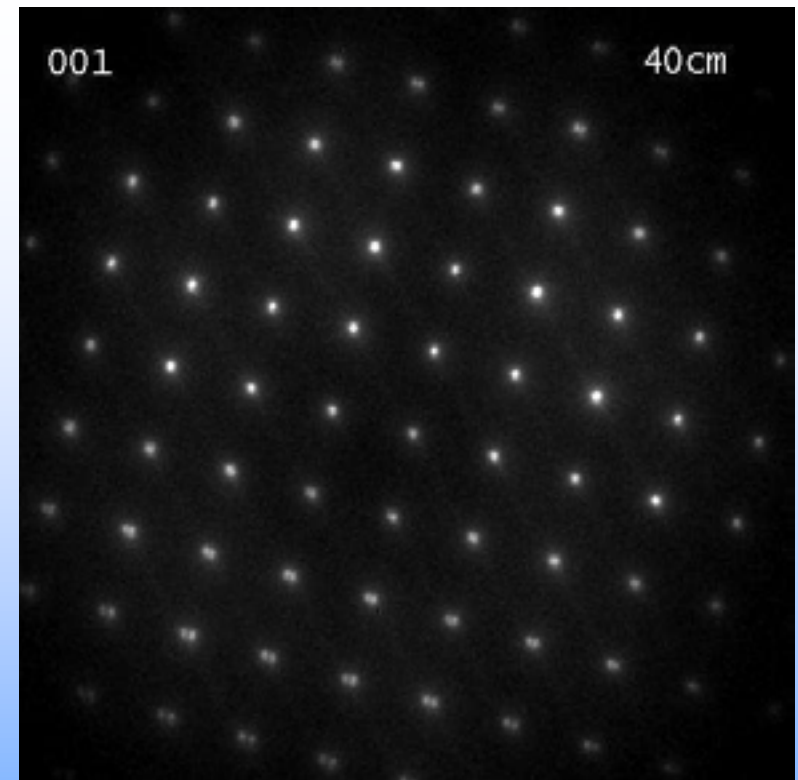
Stress relaxation by twinning



TEM micrographs of $\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$ perovskite grains showing a complex configuration of twins

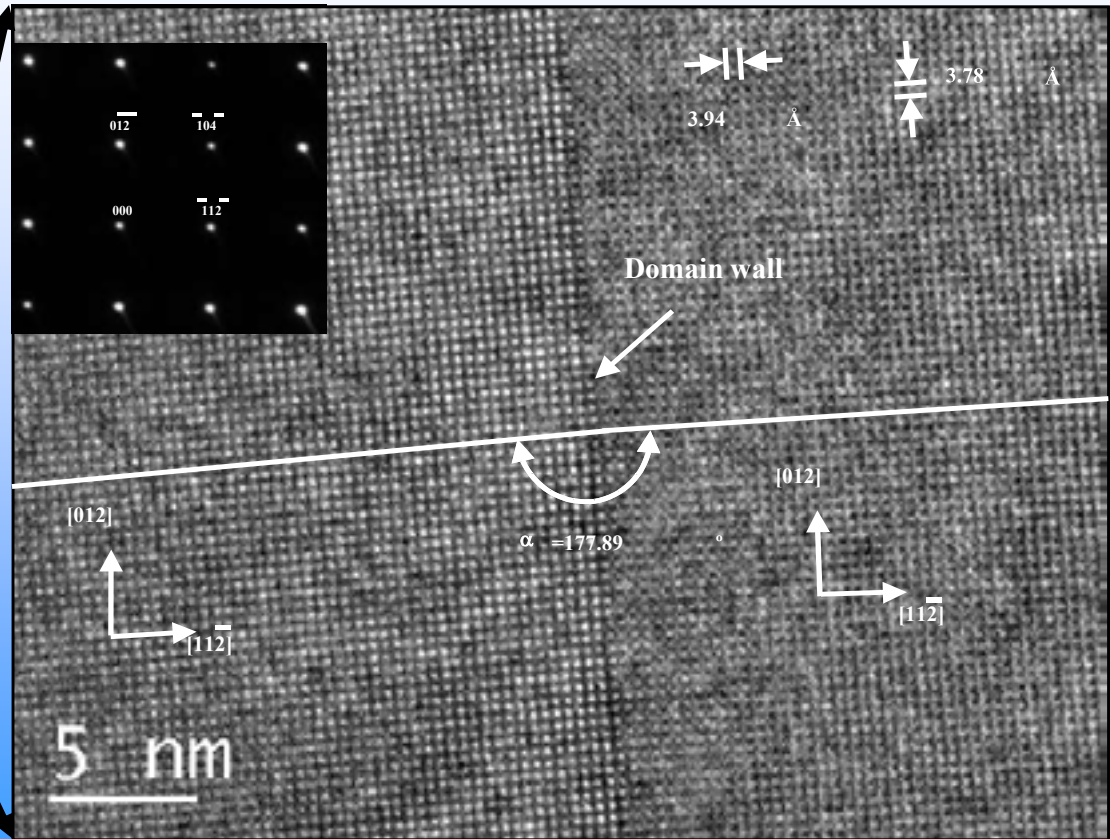
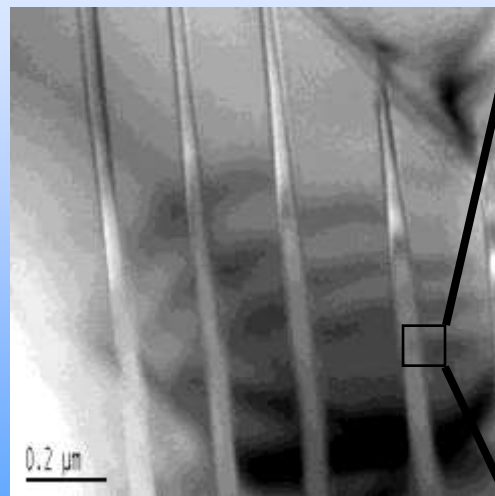


TEM micrograph showing a configuration of twins in LaCoO_3

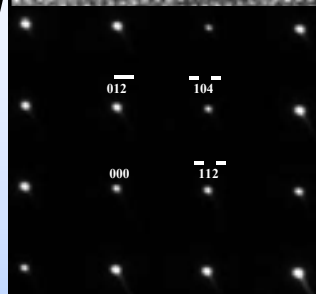


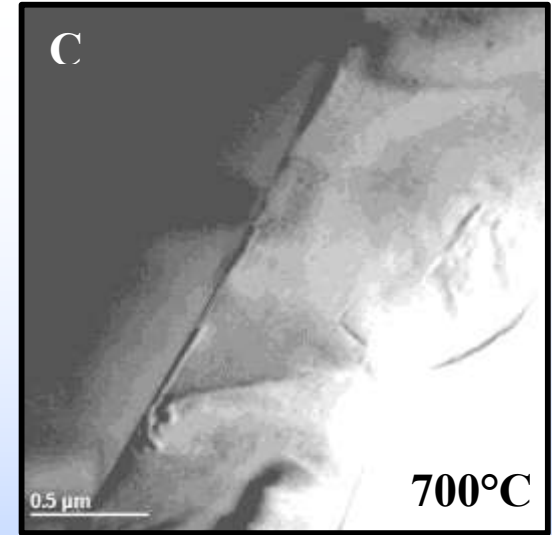
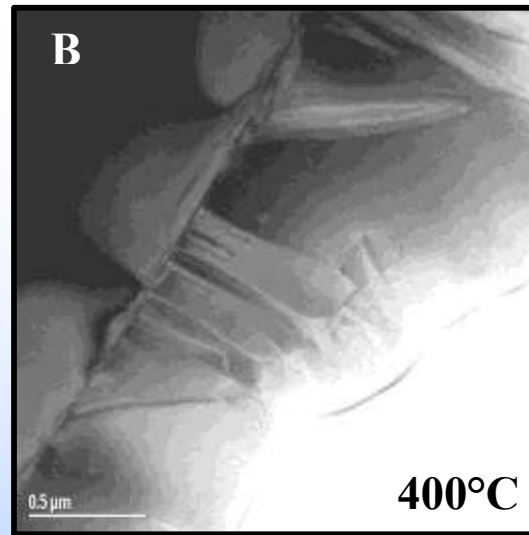
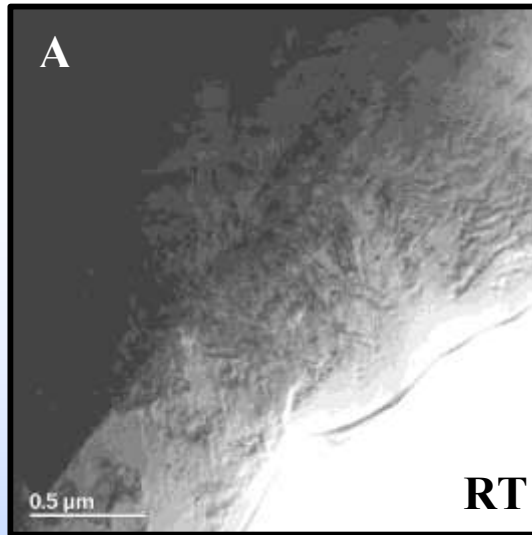
Electron diffraction pattern exhibits splitting of the spots. The magnitude of the splitting increases with the distance from the center of the pattern

Domain wall in LaCoO_3 .

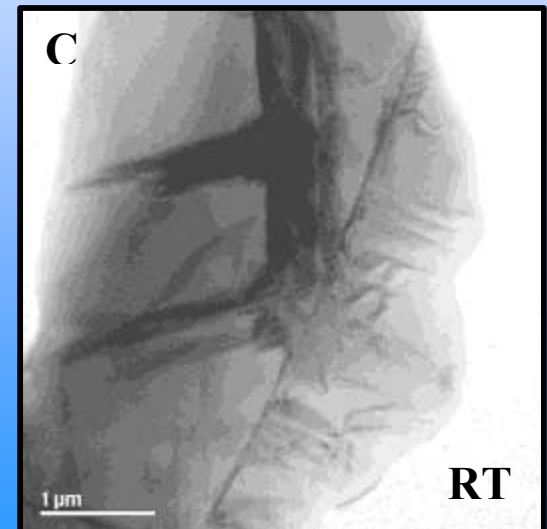
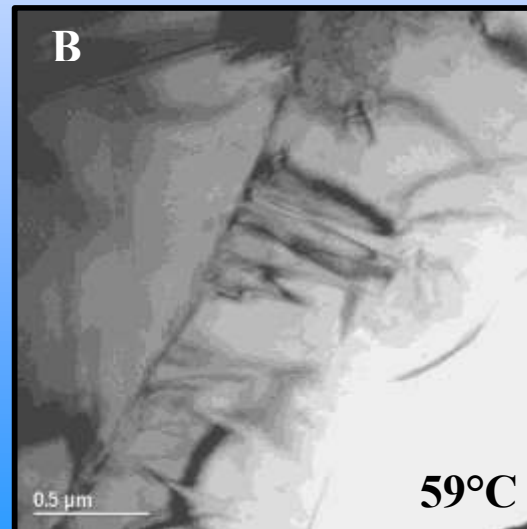
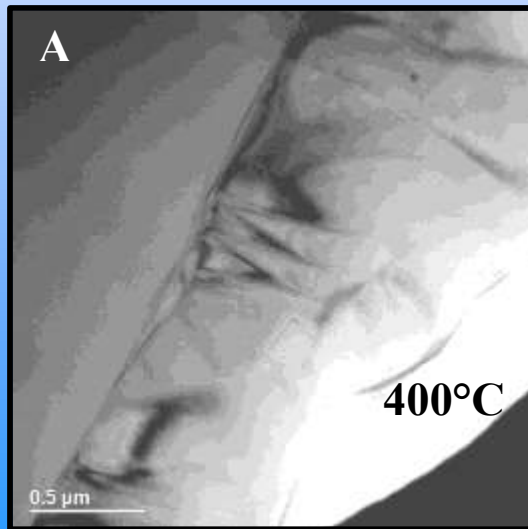


No splitting of diffraction spots in the SAD pattern has been observed

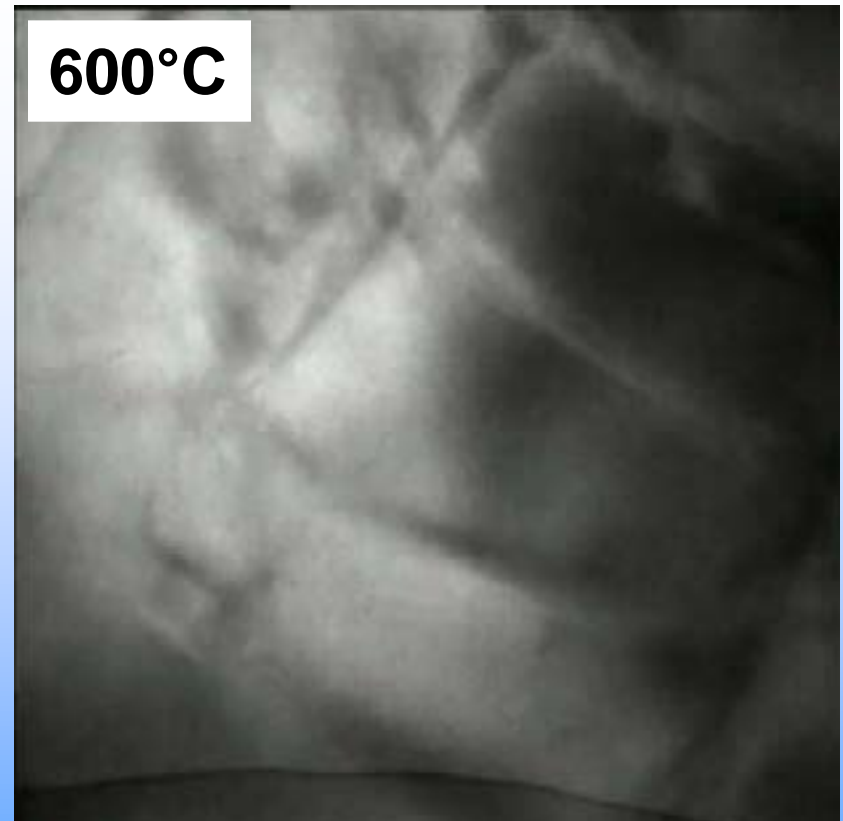
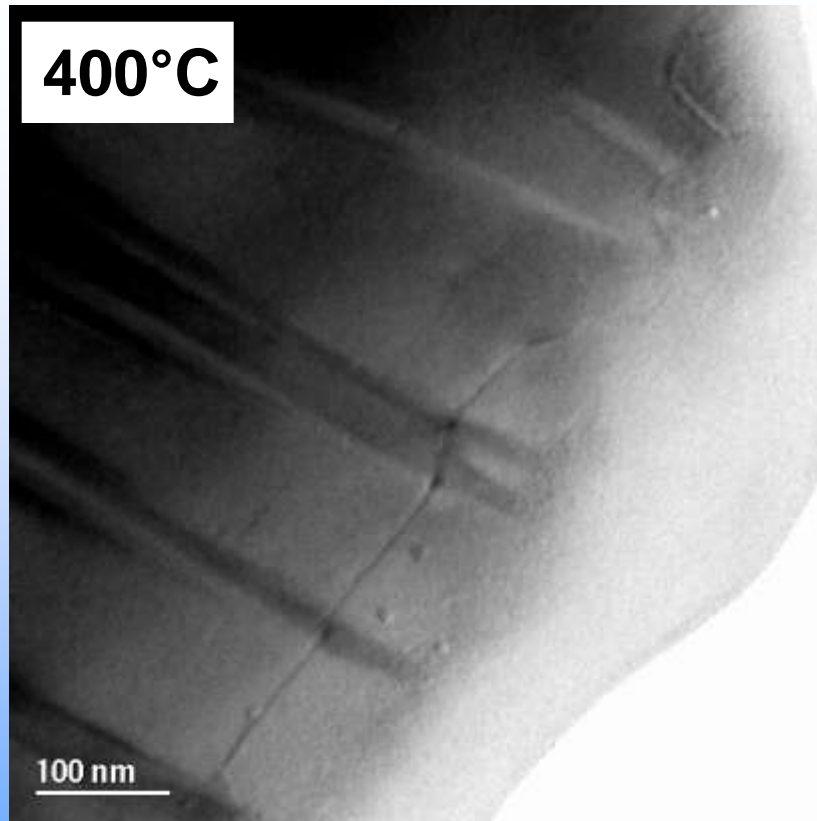




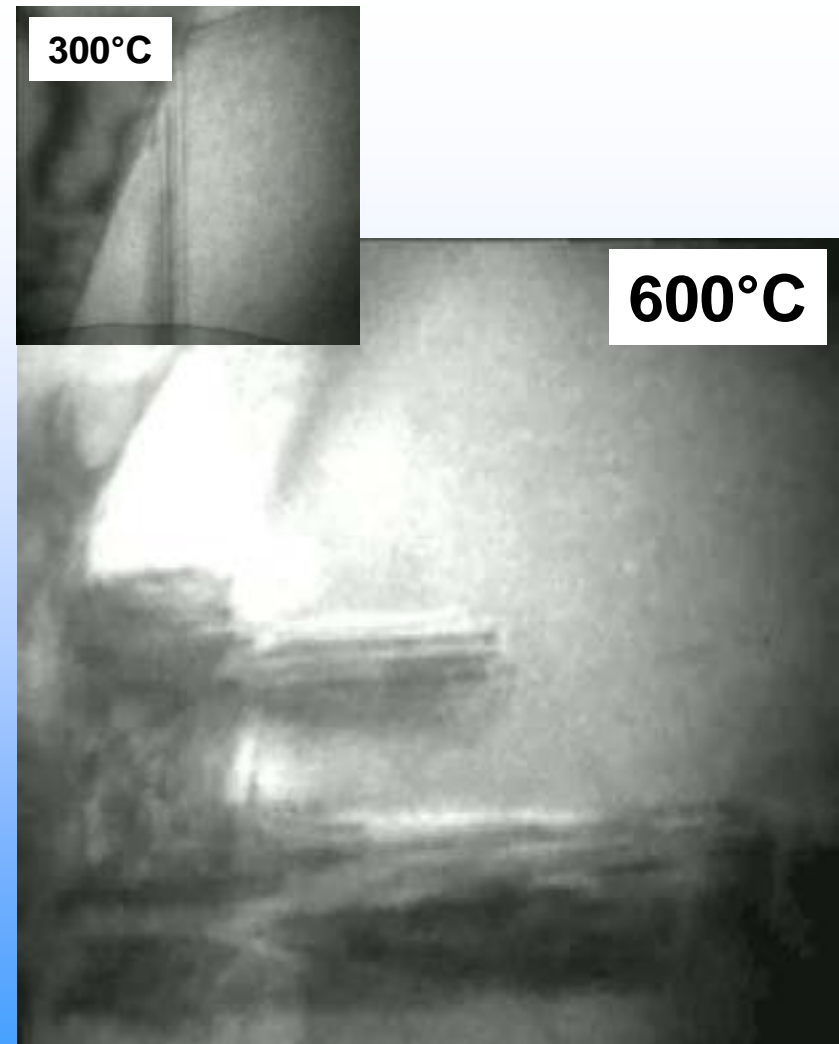
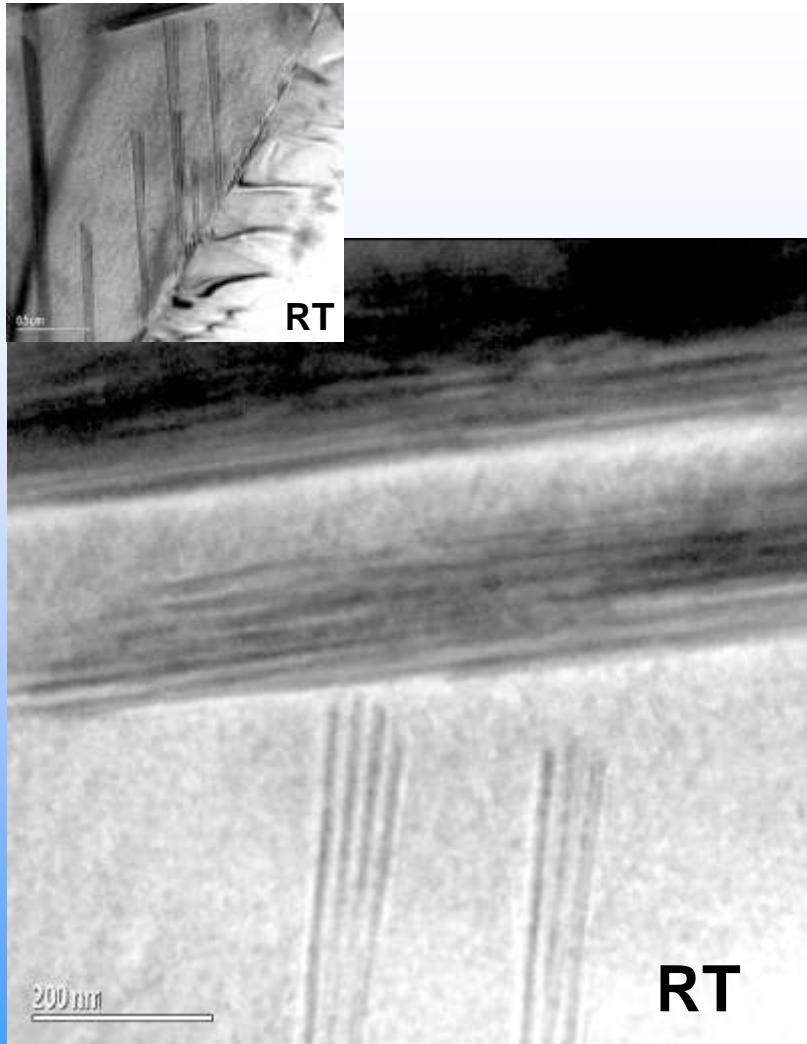
The micrographs showing detwinning upon heating in $\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$



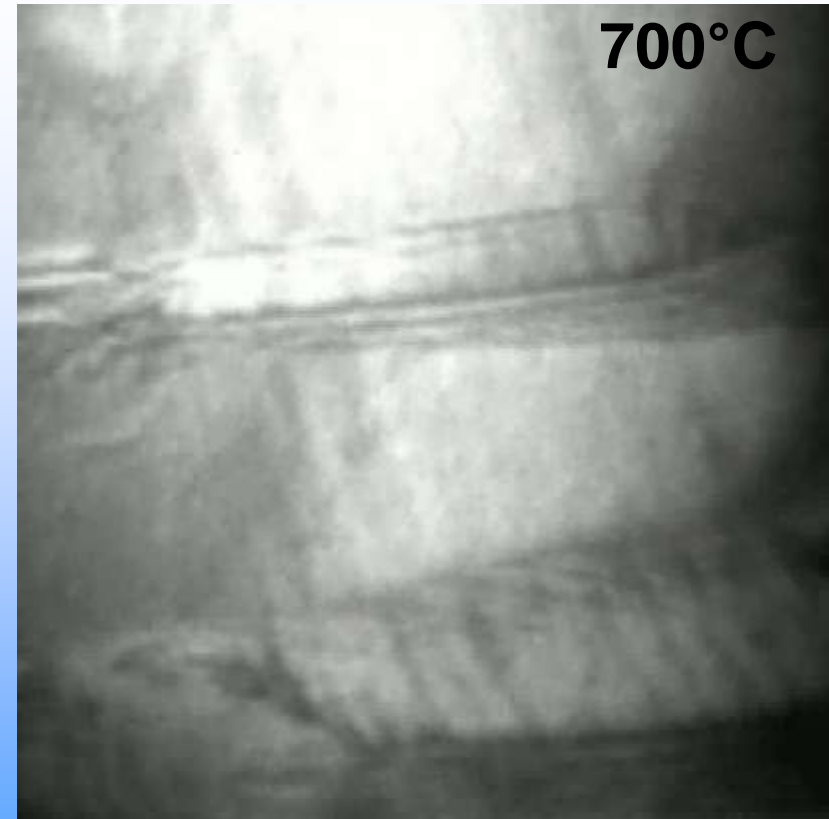
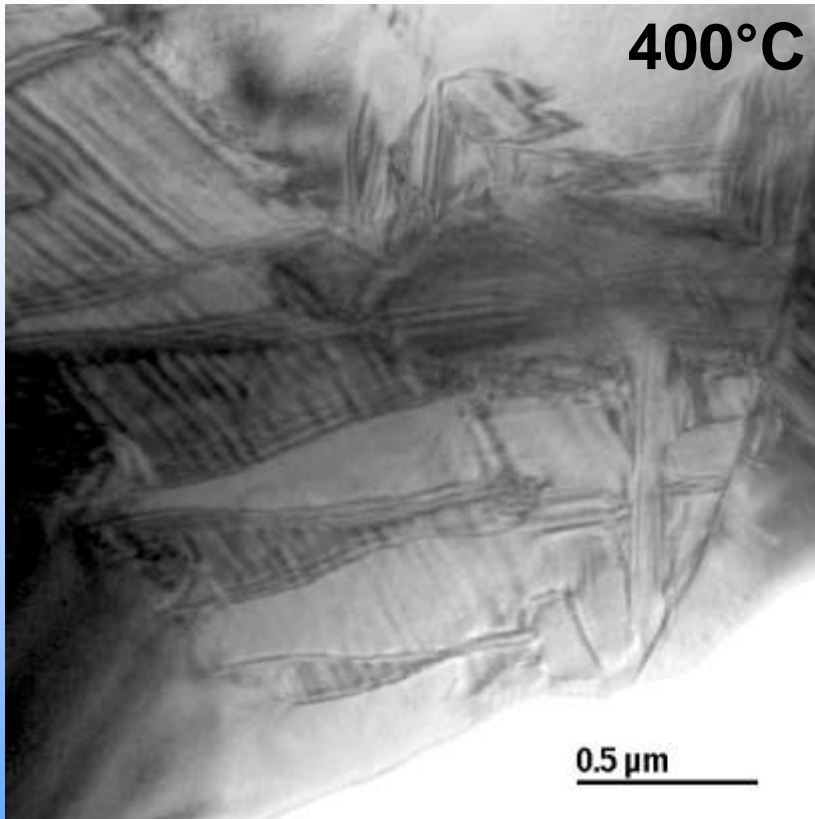
The micrographs showing reappearance of twins during cooling



**Twins in $\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$ perovskite at 400°C.
A dislocation is observed near the twins' tips.
The dislocation serves as a strong pinning point delaying
disappearance of the twins during heating.**



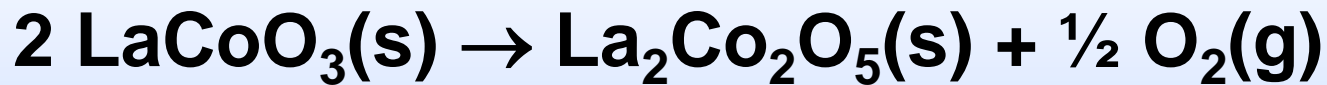
Stacking faults in $\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$ perovskite



Detwinning process during heating of $\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$

LaCoO_{2.5} structure

LaCoO₃ perovskite undergoes topotactic reduction to vacancy-ordered phase at high temperatures in reducing atmosphere

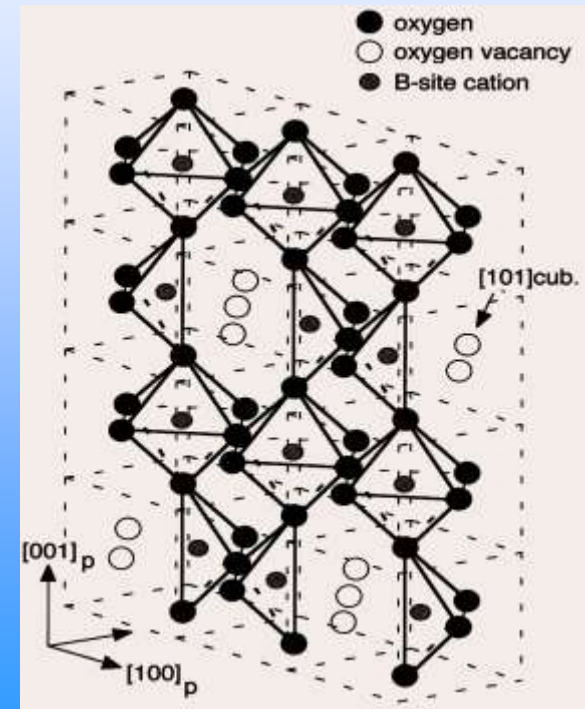


Orthorhombic brownmillerite structure

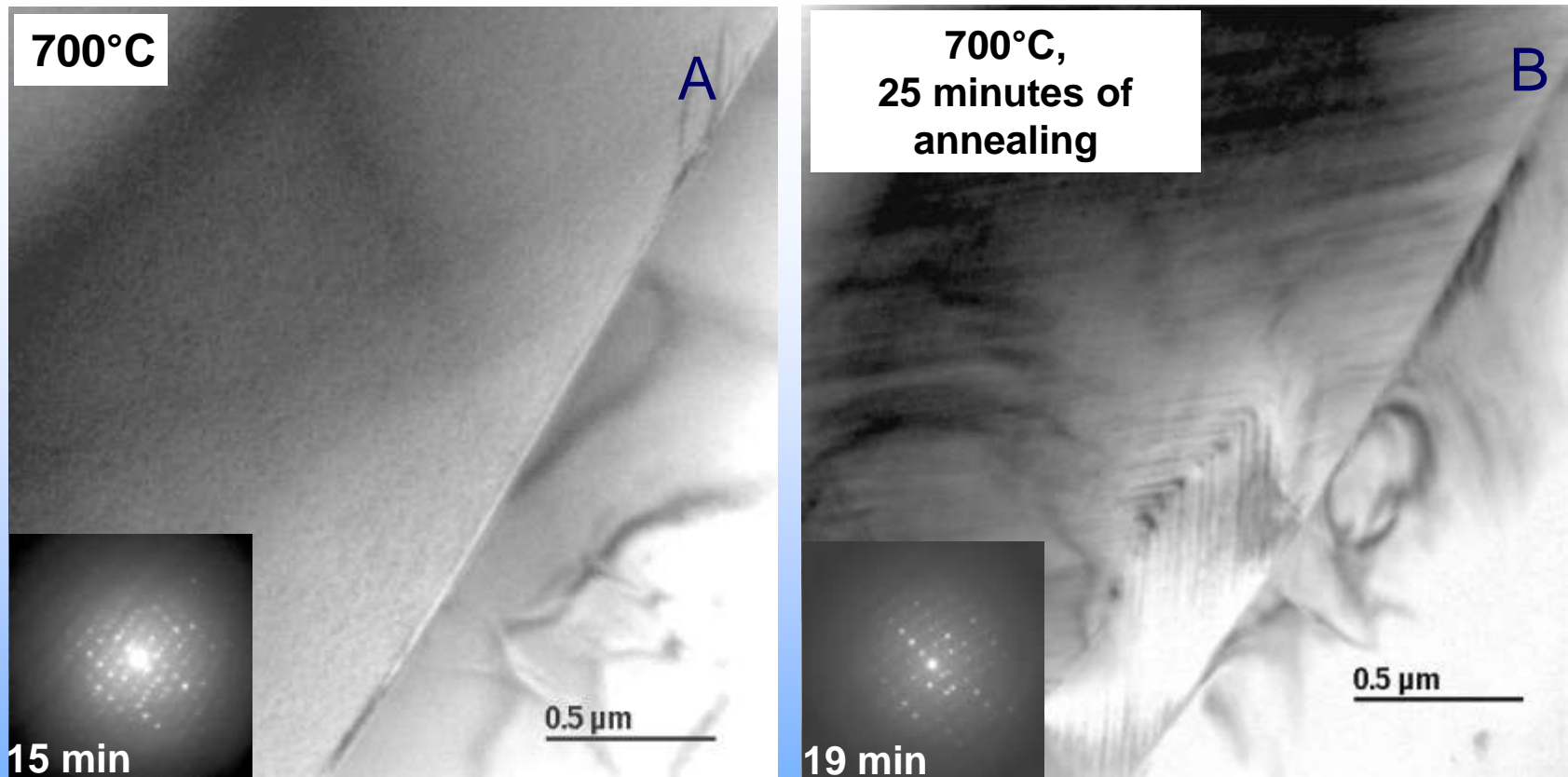
Perovskite type structure

Alternate layers of Co in octahedral and tetrahedral sites

Every second row of oxygen atoms is removed from alternate planes



Stemmer, S., Sane, A., Browning, N.D., Mazanec, T.
Solid State Ionics, 130, 71-80, 2000

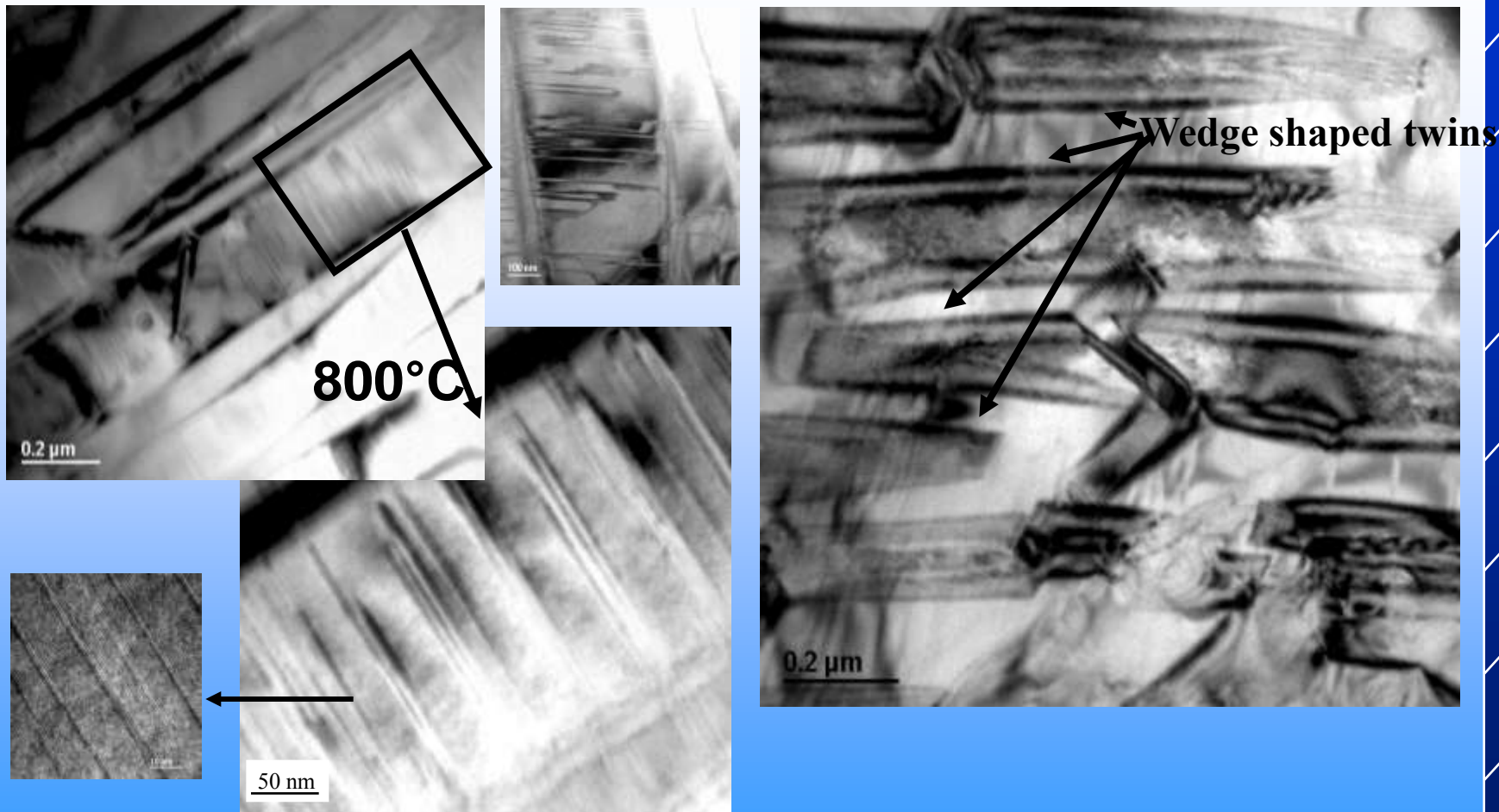


TEM micrographs of $\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$ perovskite at 700°C.

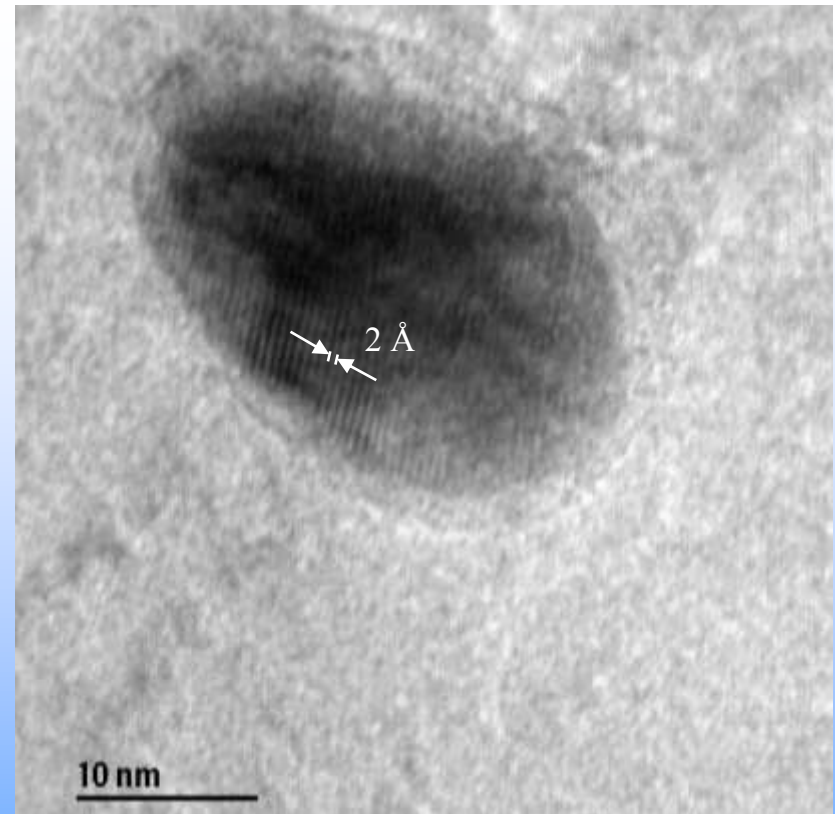
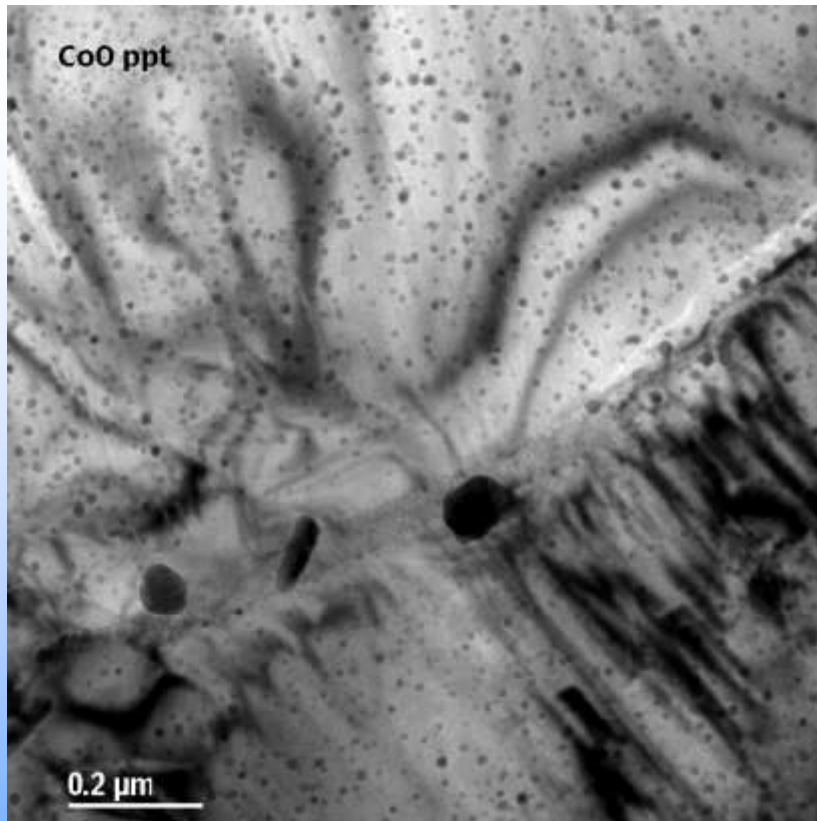
A) Micrograph taken immediately after heating. A single domain grain can be clearly seen.

B) Micrograph taken after 25 min of annealing at 700°C

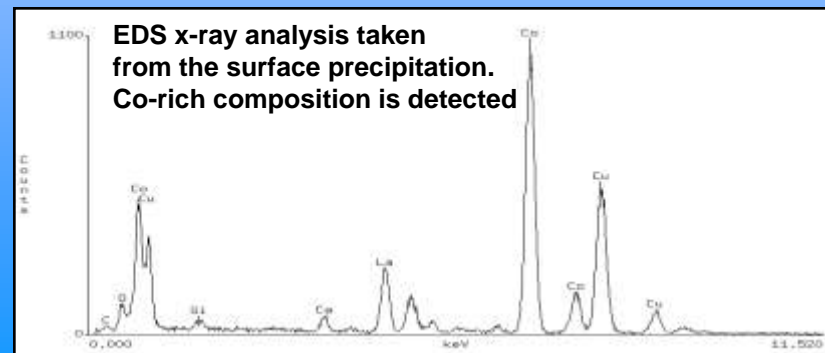
Twins in $\text{La}_{0.8}\text{Ca}_{0.2}\text{CoO}_3$



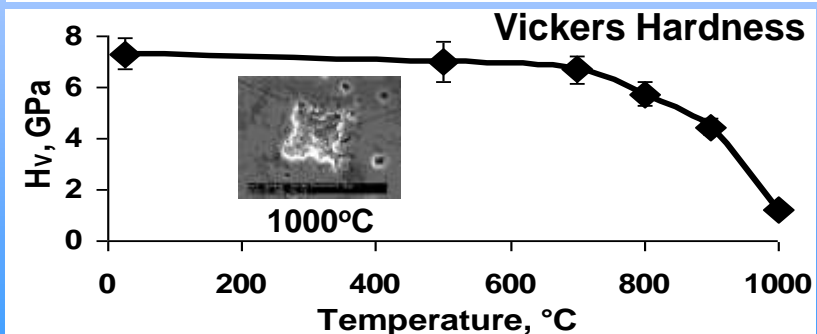
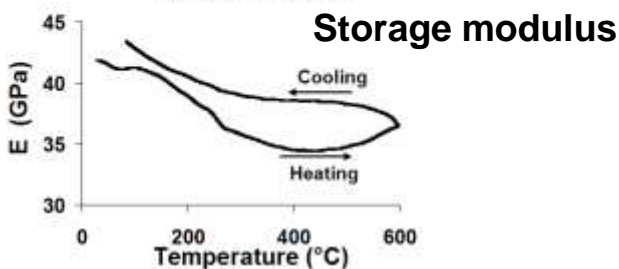
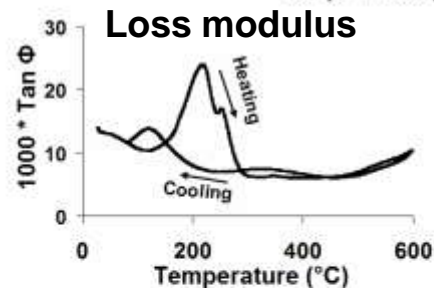
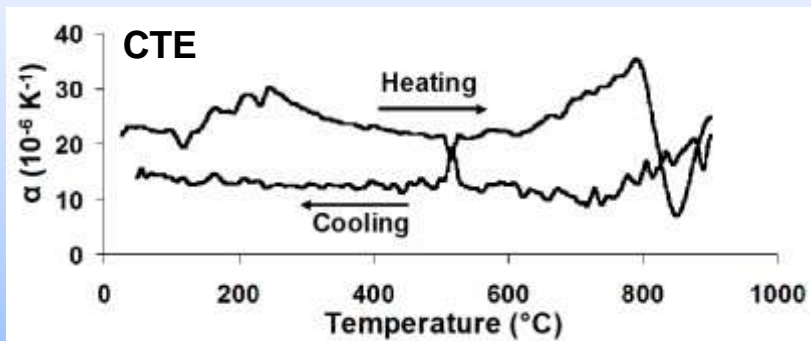
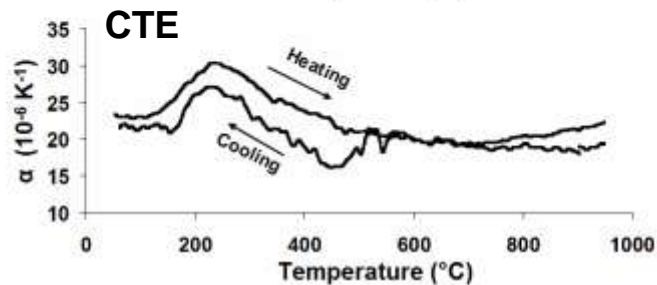
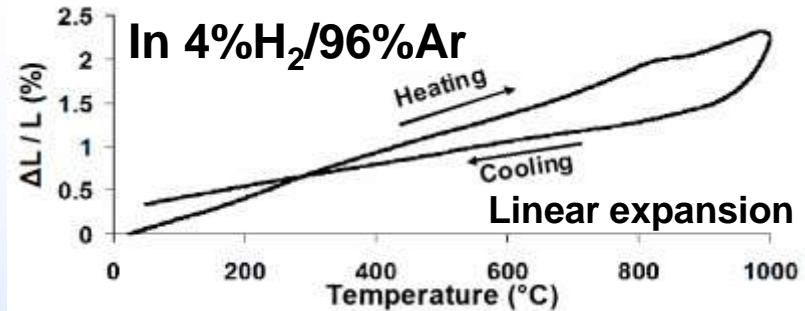
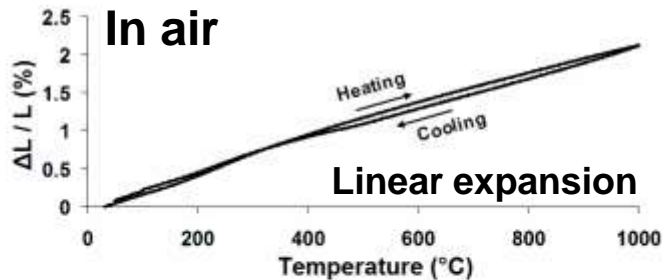
Wedge shaped twins along with short parallel to each other domains are stable up to 850°C in $\text{La}_{0.8}\text{Ca}_{0.2}\text{CoO}_3$ perovskite



Co oxide surface precipitations during annealing of $\text{La}_{0.8}\text{Ca}_{0.2}\text{CoO}_3$ for 1.5 hours at 800°C . Precipitations are favored along the grain boundaries.

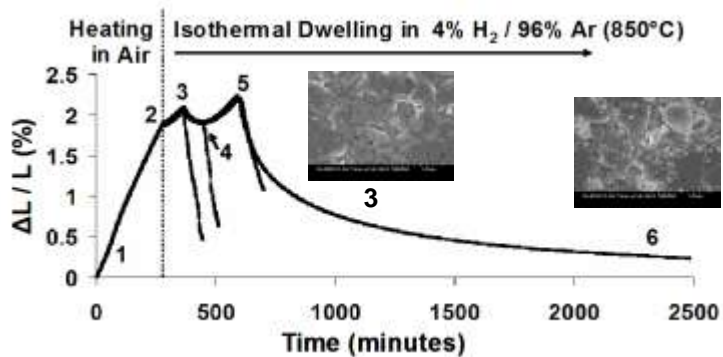


LaCoO₃ in oxidizing and reducing environment

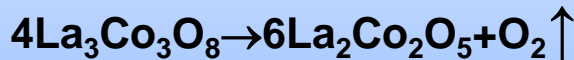


LaCoO₃ is not stable in reducing atmosphere with the most significant changes starting to occur at 800-900°C

LaCoO₃ reduction steps



#1



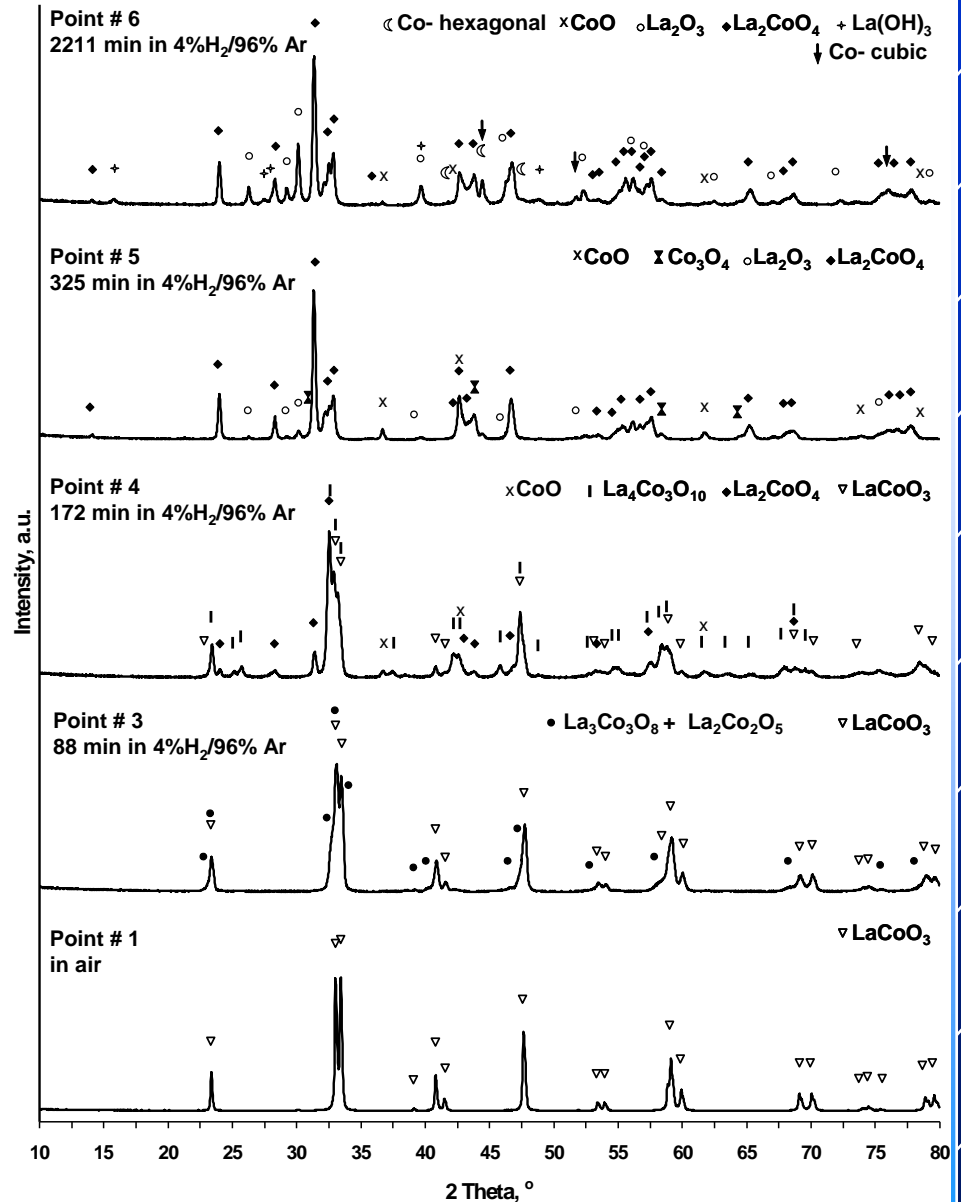
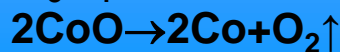
#2



#3

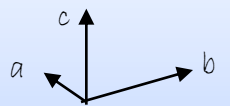


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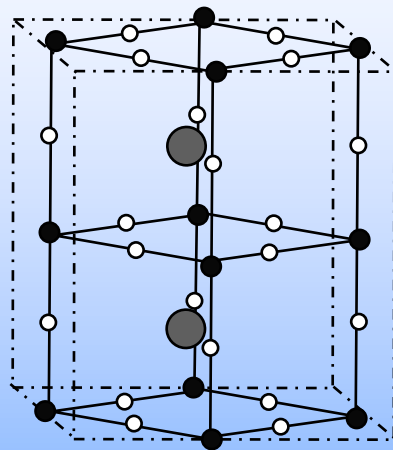


LaGaO₃ Crystal Structure

Orthorhombic



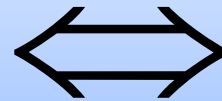
$$\begin{aligned} a &= 5.520 \text{ \AA} \\ b &= 5.490 \text{ \AA} \\ c &= 7.770 \text{ \AA} \end{aligned}$$



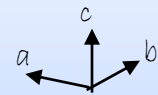
Room temperature



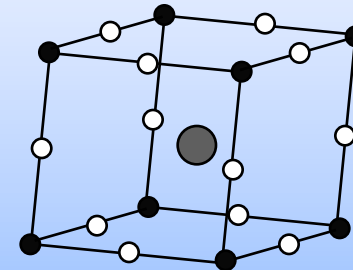
145°C



Rhombohedral



$$\begin{aligned} a &= 3.889 \text{ \AA} \\ \alpha &= 89.50^\circ \end{aligned}$$



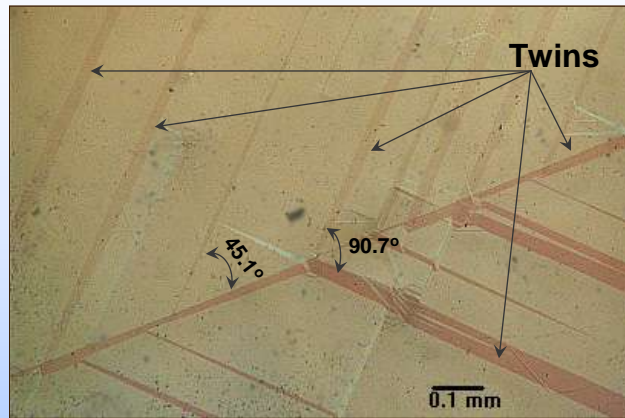
The first-order transition orthorhombic \leftrightarrow rhombohedral @ $\sim 145^\circ\text{C}$ \rightarrow leads to twinning of LaGaO₃ crystals \rightarrow change of volume ⁽²⁾

(1) I.K.Bidikin, I.M.S.k., A.M. Balbashov, A.V. Kazansky, *Twinning of LaGaO₃ Single Crystals*. J. Appl. Cryst., 1993. **26**: p. 71-76

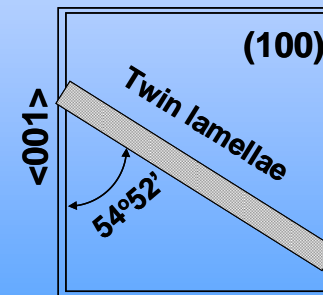
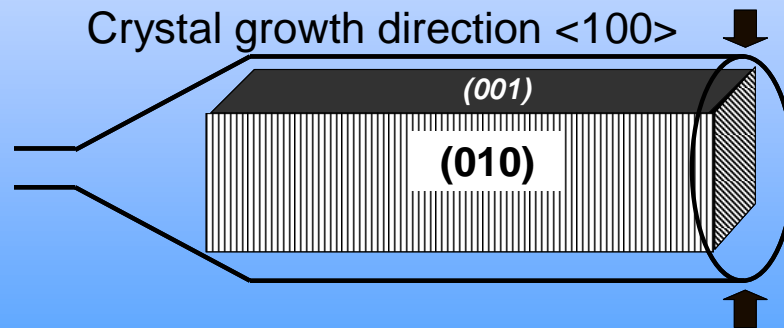
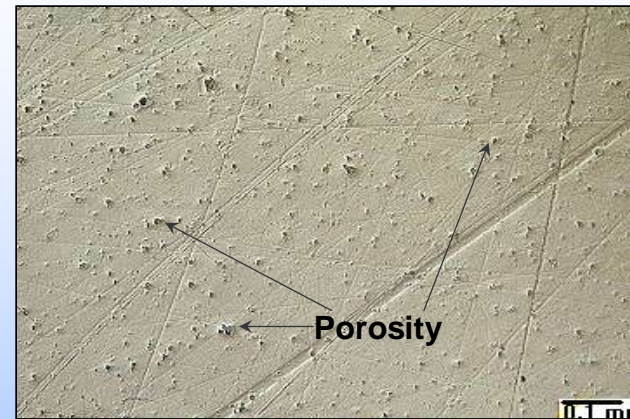
(2) A.N.Morozov, O.Y.M., N.M.Ponomarev, *Real Structure of single crystals of LaGaO₃ grown by the Czochralski method. 1. X-ray diffraction and X-ray topographic methods*. Kristallografiya, 1993. **38**: p. 149-159

LaGaO₃ single crystals preparation

LaGaO₃ single crystal in the [100] plane

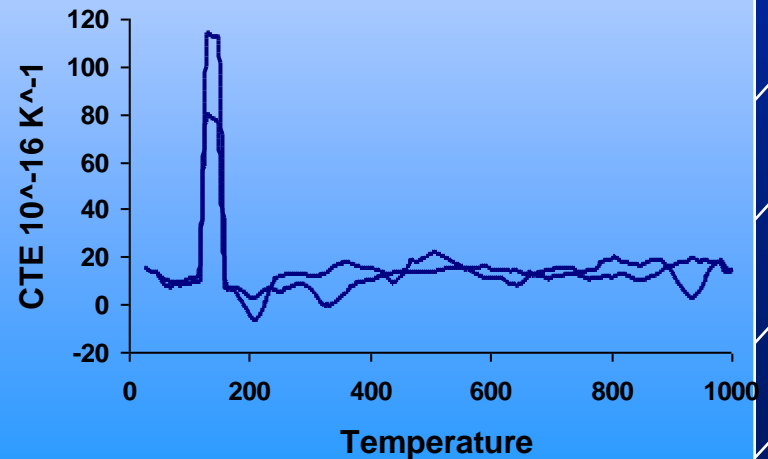
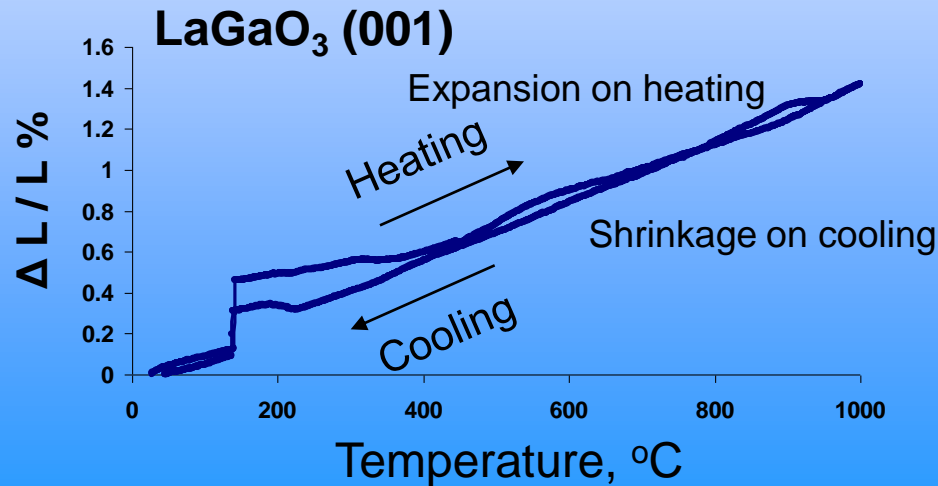
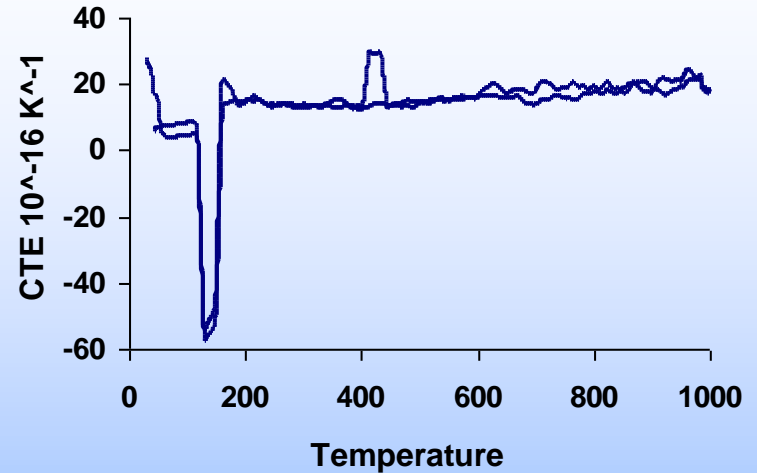
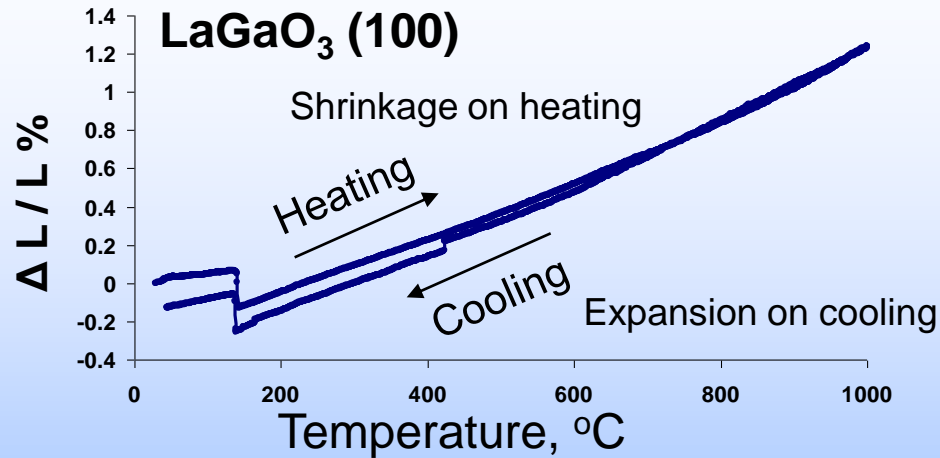


LaGaO₃ single crystal in the [001] plane



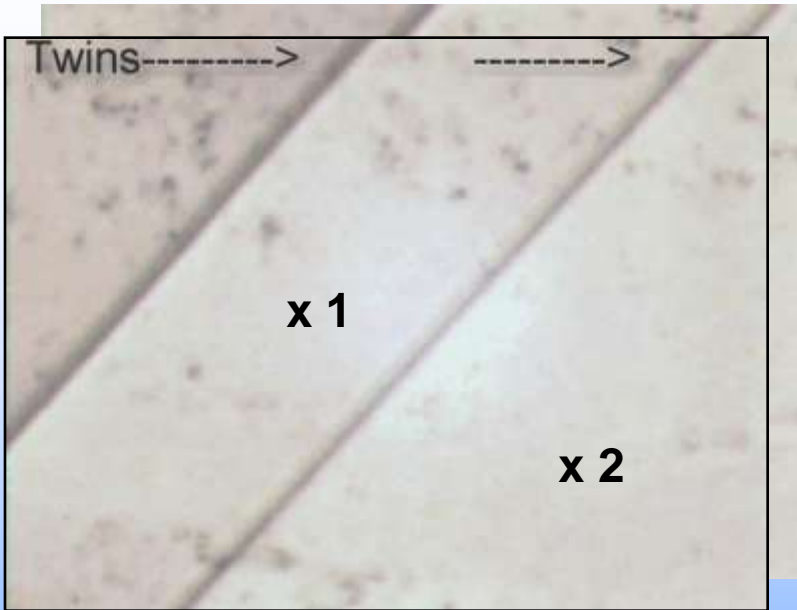
- Single-crystals of LaGaO₃ were grown by Czochralski technique
- LaGaO₃ (001) subjected to detwinning
- C. Klementz, University of Central Florida produced the crystals

LaGaO₃: Volume Change during Phase transition



Overall volume change in the phase transition < 0.01%

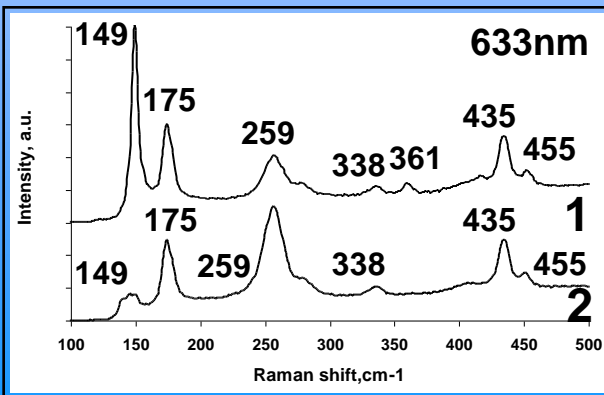
Optical microscopy: Heating of (100) LaGaO₃



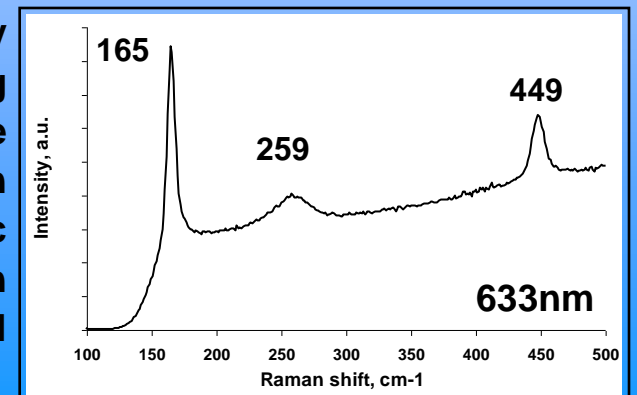
Heating 143-143.5°C



Cooling ~144°C

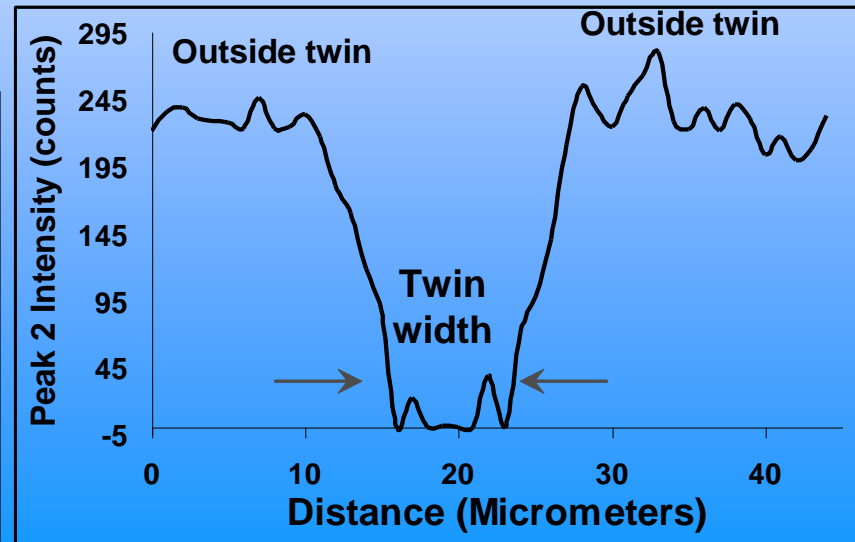
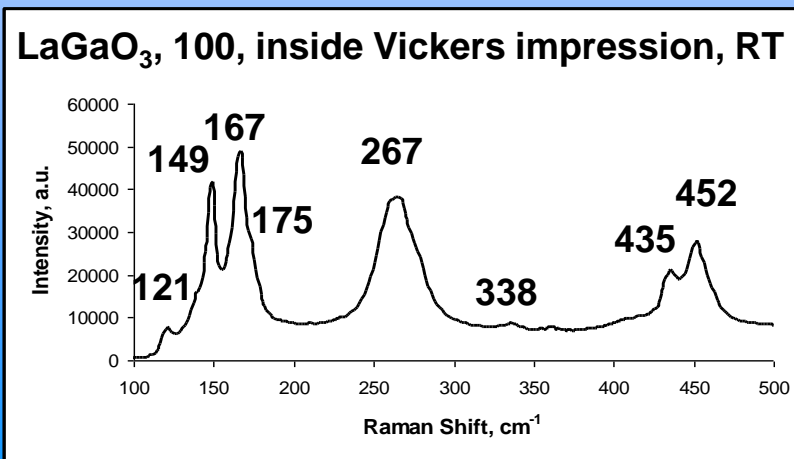
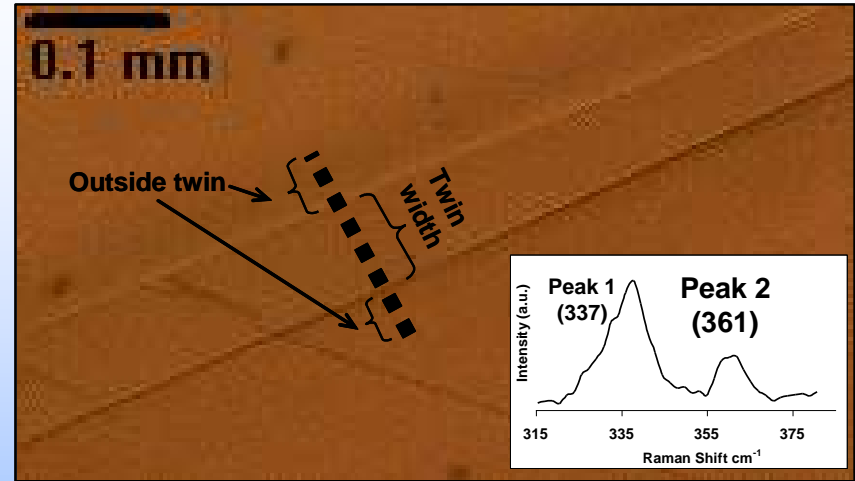
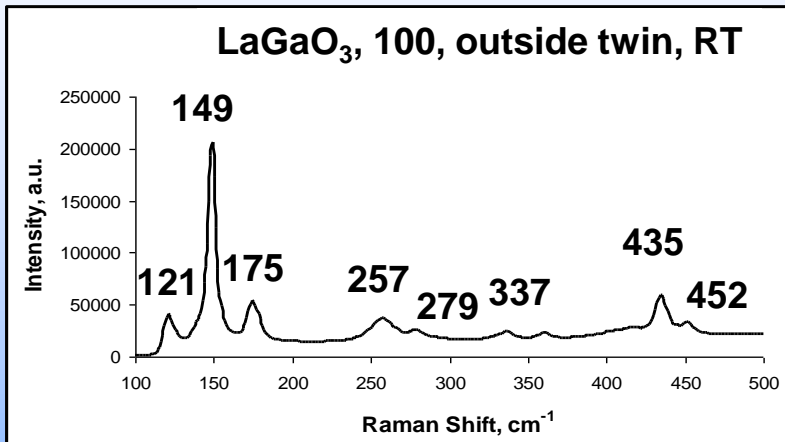


Raman spectroscopy allows distinguishing the two phases. There are 24 allowed Raman modes in orthorhombic LaGaO₃ and 5 Raman modes in rhombohedral LaGaO₃.



Raman spectra of (100) LaGaO_3

Intensity map of 361cm^{-1} Raman band



Conclusions

- Twins, domain walls, stacking faults and dislocations were observed in LaCoO_3 based perovskites.
- The mobility of domains was recorded *in-situ* by high temperature TEM in the $\text{La}_{0.6}\text{Ca}_{0.4}\text{CoO}_3$ perovskite.
- The detwinning process during the heating and reappearance of domains during cooling were observed as a result of a ferroelastic phase transition.
- Annealing of the LaCoO_3 based specimens at high temperatures in vacuum leads to progressive transitions accompanied by nucleation and movement of new stacking faults. This can be explained by the formation of oxygen deficient brownmillerite-type $\text{LaCoO}_{2.5}$ structure.
- LaCoO_3 is not stable in reducing environment and can be easily reduced up to metallic cobalt.
- The orthorhombic to rhombohedral phase transition in LaGaO_3 occurs at $\sim 145^\circ\text{C}$. The significant thermal expansion/contraction occurs in the material during heating/cooling cycles.